

Scan widths were $(1.20 + 0.35 \tan\theta)^\circ$ in ω , with a background/scan time ratio of 0.5. The data were corrected for Lorentz and polarization effects.

The Laue group assignment, systematic absences and intensity statistics consistent with centrosymmetry indicated space group $C2/c$ (No. 15); since refinement proceeded well, it was adopted. Fourier difference methods were used to locate the H-atom positions. Full-matrix least-squares refinement was performed. In later stages of the refinement, aromatic-ring-H atoms were made canonical: C—H distance = 0.98 Å; $U_{iso} = 1.2U_{eq}$ of the attached atom. Atoms H9A and H9B (on C9A and C9B) and carboxyl-H atom H_B were refined isotropically.

The A molecule carboxyl-O atoms were found to be disordered over two sites, labelled A and A*, with (refined and final) occupancies of 0.66 (2) and 0.34 (2), respectively. Following refinement with all but the two partially occupied carboxylic-H atoms, H_A and H_A^{*}, included in the model, they were assigned fixed coordinates from the resulting Fourier difference map and were assigned a fixed U_{iso} very slightly larger than that of the refined B molecule carboxylic-H atom, H_B.

The maximum effect of extinction was 4.7% of F_o for 004. The maximum peak in the final difference map occurred ~ 1.2 Å from O1A* and O2A, but was not suitably oriented to be assigned as a potential partial-occupancy carboxylic-H atom; the minimum peak occurred near the center of one of the benzenoid rings of the B molecule.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1988). Cell refinement: *MSC/AFC Diffractometer Control Software*. Data reduction: *TEXSAN* (Molecular Structure Corporation, 1989). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *TEXSAN*. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *TEXSAN*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: BK1245). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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A New Conformationally Restricted Aspartic Acid Analogue with a Cyclohexanone Skeleton

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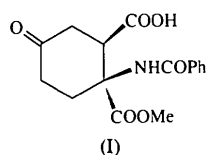
Abstract

In the compound (1*R*,2*S*)-2-benzamido-2-methoxy-carbonyl-5-oxocyclohexane-1-carboxylic acid, C₁₆H₁₇NO₆, the cyclohexanone ring adopts a distorted chair conformation. The carboxylic acid and the methyl ester groups occupy the axial positions, while the benzamido group is equatorial. The values determined for the torsion angles about the N—C^α(φ) and C^α—CO(ψ) bonds correspond to a semi-extended conformation for the amino acid residue. The crystal structure is stabilized by two intermolecular hydrogen bonds (O—H \cdots O and N—H \cdots O) involving the carboxylic acid, the benzamido and the methyl ester groups.

Comment

The use of uncommon amino acid residues in the synthesis of peptides with restricted conformational flexibility has acquired increasing importance in the design of specifically folded analogues of biologically active

peptides (Liskamp, 1994; Gante, 1994). In this context, we have developed a new methodology for the stereoselective synthesis of new non-proteinogenic conformationally restricted α -amino acids that involves the use of (*Z*)-4-[(*S*)-2,2-dimethyl-1,3-dioxolan-4-yl-methylidene]-2-phenyl-5(*4H*)-oxazolone as a dienophile in Diels–Alder reactions with different dienes (Buñuel, Cativiela & Díaz-de-Villegas, 1994, 1995, 1996; Buñuel, Cativiela, Díaz-de-Villegas & García, 1994) and, as part of our investigations of the conformational properties of some of these interesting compounds (Buñuel, Cativiela, Díaz-de-Villegas & Gálvez, 1996*a,b*), we describe here the crystal and molecular structure of a new conformationally restricted aspartic acid analogue with a cyclohexanone skeleton, (I).



A perspective view of (I) with the numbering scheme used in this analysis is shown in Fig. 1. The cyclohexanone ring adopts a distorted chair conformation [the Cremer & Pople (1975) puckering parameters are $Q = 0.497(2)$ Å and $\theta = 10.1(2)^\circ$] in which the C1 and C4 atoms are displaced from the plane defined by the other four atoms by 0.592(2) and $-0.547(2)$ Å, respectively. As for the torsion angles relating the cyclohexane ring to the substituents, the (+*sc*, -*sc*) conformation (axial disposition) is adopted by the carboxylic acid and methyl ester groups, while the (*ap*, *ap*) conformation (equatorial disposition) is adopted by the benzamido group.

The methyl ester group attached to atom C1 is planar and adopts a nearly eclipsed conformation with respect to the C1—C2 bond of the cyclohexane ring [the torsion angle C2—C1—C7—O3 is $-12.0(2)^\circ$].

The angle between the average planes of the amide and phenyl groups in the benzamido moiety is $33.8(1)^\circ$.

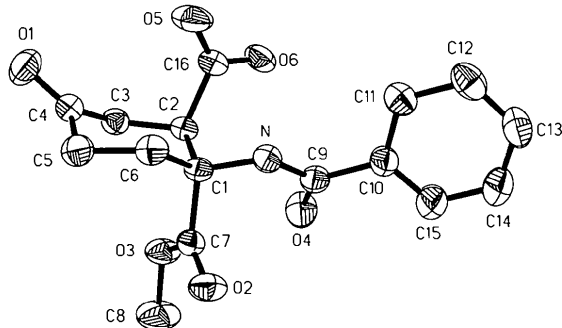


Fig. 1. The molecular structure of (I) showing 50% probability displacement ellipsoids. H atoms have been omitted for clarity.

The amide linkage is found in the usual *trans* conformation [the torsion angle C1—N—C9—C10 is $174.8(2)^\circ$] and adopts an antiperiplanar conformation with respect to the C1—C6 bond of the cyclohexane ring [the torsion angle C9—N—C1—C6 is $169.7(2)^\circ$]. The torsion angle C1—C2—C16—O5 of $-72.9(2)^\circ$ indicates that the axial carboxylic acid group is conveniently orientated in a conformation that avoids an eclipsing of the C2—C3 and C16—O5 bonds.

The amino acid residue adopts a semi-extended conformation; the values of the backbone torsion angles C9—N—C1—C7 [$\varphi = 53.5(2)^\circ$] and N—C1—C7—O3 [$\psi = -132.4(2)^\circ$] (IUPAC–IUB Commission on Biochemical Nomenclature, 1970) fall in the F^* region of the conformational map (Zimmerman, Pottle, Nemethy & Scheraga, 1977). The values of the torsion angles N—C1—C2—C3 [$\chi^{11} = -163.9(1)^\circ$] and N—C1—C6—C5 [$\chi^{12} = 172.3(2)^\circ$] relating the peptide chain to the cyclohexane ring reflect an (*ap*, *ap*) conformation.

In the crystal, the molecules form an infinite layer perpendicular to the crystallographic *z* axis (Fig. 2), characterized by two different intermolecular hydrogen bonds, *i.e.* an (amide)N—H...O=C(methyl ester) and an (acid)O—H...O=C(amide) hydrogen bond [N...O2ⁱ 3.003(2), H...O2ⁱ 2.09 Å and N—H...O2ⁱ 176°; O6...O4ⁱⁱ 2.597(2), HO...O4ⁱⁱ 1.794(2) Å and O6—HO...O4ⁱⁱ 150°; symmetry codes: (i) $1-x, -\frac{1}{2}+y, \frac{1}{2}-z$; (ii) $2-x, -\frac{1}{2}+y, \frac{1}{2}-z$]. The layers are held together by van der Waals forces between the phenyl groups.

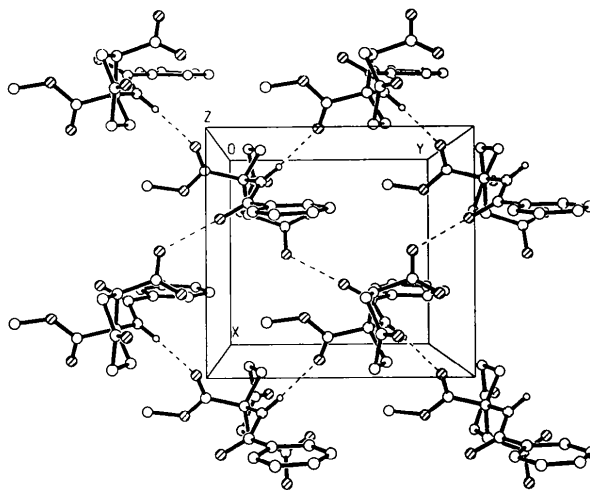


Fig. 2. Packing diagram viewed down the crystallographic *z* axis; O—H...O and N—H...O hydrogen bonds are shown as dashed lines.

Experimental

The title compound was prepared according to a procedure described by Buñuel, Cativiela & Díaz-de-Villegas (1996).

Crystals were obtained by slow evaporation from a hexane solution.

Crystal data

$C_{16}H_{17}NO_6$

$M_r = 319.31$

Orthorhombic

$P2_12_12_1$

$a = 8.707(1) \text{ \AA}$

$b = 9.313(1) \text{ \AA}$

$c = 19.869(1) \text{ \AA}$

$V = 1611.1(3) \text{ \AA}^3$

$Z = 4$

$D_x = 1.316 \text{ Mg m}^{-3}$

D_m not measured

Mo $K\alpha$ radiation

$\lambda = 0.71069 \text{ \AA}$

Cell parameters from 32 reflections

$\theta = 3.80\text{--}12.49^\circ$

$\mu = 0.102 \text{ mm}^{-1}$

$T = 293(2) \text{ K}$

Prism

$0.68 \times 0.60 \times 0.38 \text{ mm}$

Colourless

Data collection

Siemens P4 diffractometer

$\theta/2\theta$ scans

Absorption correction:

none

2219 measured reflections

2042 independent reflections

1917 observed reflections

$[I > 2\sigma(I)]$

$R_{int} = 0.0169$

$\theta_{max} = 25^\circ$

$h = -1 \rightarrow 10$

$k = -1 \rightarrow 11$

$l = -1 \rightarrow 23$

3 standard reflections

monitored every 97

reflections

intensity decay: none

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.0321$

$wR(F^2) = 0.0871$

$S = 1.063$

2042 reflections

209 parameters

Only U_{iso} refined for each H atom

$w = 1/[\sigma^2(F_o^2) + (0.05P)^2 + 0.1993P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.001$

$\Delta\rho_{max} = 0.157 \text{ e \AA}^{-3}$

$\Delta\rho_{min} = -0.167 \text{ e \AA}^{-3}$

Extinction correction: none

Atomic scattering factors

from *International Tables for Crystallography* (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Absolute configuration:

assigned on the basis of the known configuration of the (Z)-4-[(S)-2,2-dimethyl-1,3-dioxolan-4-yl-methylidene]-2-phenyl-5(4H)-oxazolone reagent

C11	0.8003 (2)	0.1169 (2)	0.15274 (11)	0.0426 (5)
C12	0.8174 (2)	0.0359 (3)	0.09484 (12)	0.0494 (5)
C13	0.8384 (4)	0.1018 (3)	0.03411 (12)	0.0707 (8)
C14	0.8482 (6)	0.2490 (4)	0.03095 (13)	0.104 (2)
C15	0.8314 (5)	0.3311 (3)	0.08858 (12)	0.0815 (11)
C16	0.8855 (2)	0.2126 (2)	0.35769 (10)	0.0363 (4)

Table 2. Selected geometric parameters (\AA , $^\circ$)

N—C9	1.342 (2)	C2—C16	1.519 (3)
N—C1	1.465 (2)	C2—C3	1.549 (3)
O1—C4	1.219 (2)	C3—C4	1.499 (3)
O2—C7	1.202 (2)	C4—C5	1.497 (3)
O3—C7	1.325 (2)	C5—C6	1.532 (3)
O3—C8	1.449 (3)	C9—C10	1.501 (3)
O4—C9	1.229 (2)	C10—C15	1.368 (3)
O5—C16	1.205 (2)	C10—C11	1.387 (3)
O6—C16	1.314 (2)	C11—C12	1.384 (3)
C1—C6	1.540 (3)	C12—C13	1.366 (4)
C1—C7	1.549 (2)	C13—C14	1.375 (4)
C1—C2	1.560 (3)	C14—C15	1.385 (4)
C9—N—C1	121.44 (15)	O2—C7—O3	125.2 (2)
C7—O3—C8	116.4 (2)	O2—C7—C1	122.0 (2)
N—C1—C6	107.03 (14)	O3—C7—C1	112.72 (15)
N—C1—C7	107.21 (14)	O4—C9—N	120.4 (2)
C6—C1—C7	108.4 (2)	O4—C9—C10	122.6 (2)
N—C1—C2	109.28 (14)	N—C9—C10	117.0 (2)
C6—C1—C2	111.83 (14)	C15—C10—C11	119.7 (2)
C7—C1—C2	112.81 (14)	C15—C10—C9	118.9 (2)
C16—C2—C3	108.9 (2)	C11—C10—C9	121.4 (2)
C16—C2—C1	108.95 (15)	C12—C11—C10	119.9 (2)
C3—C2—C1	113.9 (2)	C13—C12—C11	120.2 (2)
C4—C3—C2	114.5 (2)	C12—C13—C14	119.8 (2)
O1—C4—C5	122.1 (2)	C13—C14—C15	120.4 (3)
O1—C4—C3	122.1 (2)	C10—C15—C14	119.9 (3)
C5—C4—C3	115.68 (15)	O5—C16—O6	124.9 (2)
C4—C5—C6	111.7 (2)	O5—C16—C2	123.9 (2)
C5—C6—C1	112.3 (2)	O6—C16—C2	111.2 (2)
C9—N—C1—C6	169.7 (2)	C7—C1—C6—C5	-72.4 (2)
C9—N—C1—C7	53.5 (2)	C8—O3—C7—C1	176.0 (2)
C6—C1—C2—C3	76.2 (2)	N—C1—C7—O3	-132.4 (2)
N—C1—C2—C3	-163.9 (1)	C2—C1—C7—O3	-12.0 (2)
C7—C1—C2—C3	77.0 (2)	C1—N—C9—C10	174.8 (2)
C16—C2—C3—C4	-80.2 (2)	C1—C2—C16—O5	-72.9 (2)
N—C1—C6—C5	172.3 (2)		

Data collection: XSCANS (Siemens, 1993). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SIR92 (Altomare *et al.*, 1992). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXTL-Plus (Sheldrick, 1989). Software used to prepare material for publication: SHELXL93. Molecular geometry calculations: PARST (Nardelli, 1983).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1251). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)

	x	y	z	U_{eq}
N	0.6925 (2)	0.3084 (2)	0.25896 (7)	0.0339 (3)
O1	0.6665 (2)	0.2918 (2)	0.52787 (7)	0.0587 (5)
O2	0.5351 (2)	0.5743 (2)	0.27031 (7)	0.0473 (4)
O3	0.7228 (2)	0.63323 (14)	0.34254 (7)	0.0475 (4)
O4	0.8562 (2)	0.4711 (2)	0.21835 (8)	0.0545 (4)
O5	0.8262 (2)	0.1110 (2)	0.38483 (9)	0.0582 (4)
O6	1.0080 (2)	0.2061 (2)	0.31920 (8)	0.0508 (4)
C1	0.6767 (2)	0.3829 (2)	0.32356 (9)	0.0302 (4)
C2	0.8285 (2)	0.3660 (2)	0.36456 (9)	0.0306 (4)
C3	0.8115 (2)	0.4022 (2)	0.44029 (9)	0.0360 (4)
C4	0.6672 (2)	0.3466 (2)	0.47215 (9)	0.0386 (4)
C5	0.5229 (2)	0.3693 (2)	0.43263 (10)	0.0430 (5)
C6	0.5389 (2)	0.3149 (2)	0.36021 (10)	0.0376 (4)
C7	0.6375 (2)	0.5416 (2)	0.30784 (9)	0.0338 (4)
C8	0.7010 (4)	0.7841 (2)	0.3276 (2)	0.0734 (9)
C9	0.7873 (2)	0.3571 (2)	0.21086 (9)	0.0367 (4)
C10	0.8052 (3)	0.2656 (2)	0.14915 (10)	0.0420 (5)

$$U_{eq} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$$

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